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Improvement of the electronic, mechanical and optical properties of cubic As-doped BN alloy for energy harvesting applications



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ARTICLE INFO

Keywords: Boron arsenide nitride Elastic properties Optical properties III-V semiconductors Optoelectronics

ABSTRACT

The electronic structure, elastic and optical properties with wide energy band gap (E_g) in mixed crystal semiconductors containing As-doped BN are reported. Density functional calculations (DFT) were performed for comprehensive detection and understanding of structural, optoelectronic, and elastic properties for mixed semiconductors in the cubic phase. When applying Vegard's law for As, BN, and mixed alloys based on BN_{1-x}As_x, where x = 0, 0.25, 0.5, 0.75, and 1, we observed that the distortion of the lattice constants was slight to different condensations. Calculations of E_g values found that the values decreased with an increase in (x) condensation. The bulk modulus, and other elastic parameters were computed and confirmed that the alloys are mechanically stable. The calculations of dielectric $\mathcal{E}(\omega)$, refractive index $n(\omega)$, higher absorption $\alpha(\omega)$ peaks, and other coefficients confirm that $BN_{1-x}As_x$ alloys are favourable for practical photovoltaics. The results confirm that mixed semiconductors have great interest in next-generation optoelectronic devices. This work provides a route for optoelectronic applications.

1. Introduction

Materials such as oxides and III-V semiconductors are necessary for optoelectronic applications. Such materials have essential properties; they are optically translucent to visible light and have considerable electrical conductivity, which makes mixed alloy semiconductors very useful in solar cells and other devices [1-7]. The crucial characteristics of optoelectronic devices based on semiconductor alloys are fundamentally pertinent in designing such device systems. The design and analysis of these devices critically depend on the values of the real and imaginary dielectric parameters (ϵ_r and ϵ_{im}), optical conductivity $\sigma(\omega)$, $n(\omega)$, and other optical properties [8-12]. The structure of III-V materials consists of a cubic mineral, sphalerite, or wurtzite based on a hexagonal structure [13]. III-boron semiconductor alloys (with As, N, and In) have recently attracted significant attention. The first motivation for studying these materials comes from their energy band gaps ranging from 1.5 to 6.5 eV and their high bonding power. Therefore, these alloys are often used in light-emitting applications and significant-temperature transistors. The GaAs_{1-x}Sb_x alloy is one of the III-V semiconductors and contains GaAs and GaSb alloys utilised in optoelectronic applications [14]. These compounds have an E_g that ranges from near 0.9–1.55 eV for the GaAs and GaSb and have high $\alpha(\omega)$ properties that are beneficial for photovoltaic systems. Another example investigated is BN_xP_{1-x} mixed with two binary alloys, BP and BN. It is found that the E_g ranges between 2.07 and 6.4 eV, and optical properties were computed. They found that $\epsilon(\omega)$, $n(\omega)$, loss-energy L(ω), extinction k(ω), reflectivity R(ω), and $\alpha(\omega)$ parameters move towards higher energies. The cubic binary boron nitride (BN) and boron arsenide (BAs) have E_g near 1.5–1.9 eV and (5.2–6.6) eV,

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https://doi.org/10.1016/j.ijleo.2023.170850 Received 20 December 2022; Received in revised form 29 March 2023; Accepted 2 April 2023 Available online 7 April 2023 0030-4026/© 2023 Elsevier GmbH. All rights reserved. respectively. These alloys were found in previous calculations to have good electrical conductivity, which is considered essential for optical devices. Both alloys have good permeability over a vast range between ultraviolet (UV) and visible. Previous studies that calculated elastic parameters showed that BN and BAs are mechanically stable. Calculations of their optical and thermodynamics parameters confirm that BN and BAs alloys are suitable for thermoelectric and optoelectronic applications due to their high absorption and large E_g . For optoelectronic systems, the active compounds are grown by forming thin films on supports. The lattice constant difference and variance in thermal parameters between the layers and supports may lead to significant stresses in the upper layer's structure. Therefore, it is indispensable to calculate their elastic parameters, which describe their response to applied pressure. This paper investigates mixed cubic $BN_{1-x}As_x$ alloys in the DFT field [15-19]. Both binary BN and BAs alloy create $BN_{1-x}As_x$. These calculations indicate that new properties can be obtained, which may reveal $BN_{1-x}As_x$ to be useful in designing optoelectronic devices. The calculations of our work prepare data for future experimental works on $BN_{1-x}As_x$.

2. Method of calculations

We applied DFT calculations using the Wien₂k code to calculate the characteristics of BAs. BN, and doping alloys [20,21]. The value of the cut-off for valence electrons was as specified 500 eV, execution of expansion for the electronic wave function in the case of Vanderbilt's ultra-soft pseudo-potential. Electronic exchange-correlation effects were characterised using Perdew-Burke-Emzerhof-(PBE) method based on a general gradient approximation (GGA) approach to explore specific properties of doping compound [22-24]. Our calculations use GGA with an applied Tran–Blaha-modified Becke–Johnson (TB-mBJ) path. A k-points of $12 \times 12 \times 12$ were selected for BAs, BN, and the doping alloy. Broydene Flectchere Goldfarbe Shenno (BFGS) algorithm was employed to evaluate atom relaxation to energies of 0.00003 eV per atom, and the entire energy concourse is 0.06 eV per Å. BAs and BN binary model after some selective As atom impurity using order selected by frequently repeated terms. The binary BN and BAs have a structure belonging to the space group 216 F-43 m while doped BN_{0.75}As_{0.25} and BN_{0.25}As_{0.75} have a structure belonging to the space group 215 P-43 m. The structure of BN0.5As0.5 belongs to the space group 115 P-4 m2. We can transform this tetragonal into the cubic group by SQS or SAE whose information description can be described in Refs. [10,25]. The BN_{1-x}As_x were modelled at (x = 0.25, 0.5, and 0.75) with ordered structures designed in terms of periodically repeated supercells with the primitive cell of BN to a $2 \times 2x$ 2 consisting of 32 atoms (B and N consist of both 16 atoms) and used the supercell to simulate As-doped systems. Structural optimization was implemented by minimizing the total energy of the cell constants and atomic positions. Charpin was used to determine the optical properties for BAs, BN, and the doping alloy using a random phase approximation (RPA) method [26]. The dielectric function for the alloy was computed by using $\epsilon_1(\omega)$ and $\epsilon_2(\omega)$ components, determined by transition frequency within the valence band (VB) and conduction band (CB) states. These relations are shown below:

$\epsilon(\omega) = \epsilon_1(\omega) + \epsilon_2(\omega)$

In computations of the elastic constants, accurate single-alloy elastic constants C_{ij} were received from the DFT of the unit cell absolute energy difference as a role of particular lattice constant distortions. When checking the accuracy of the C_{ij} numerical evaluation of the materials, we used two sets of distortions for the ground state unit cell, developed by Charpin and integrated it in the Wien₂k package [26]. In cubic symmetry, only three elastic parameters (C_{11} , C_{12} , and C_{44}) were applied. The C_{11} determines the stiffness against principal strains as [27-34]:

$$C_{11}$$
- C_{12} > 0; C_{11} > 0; C_{11} 2 + C_{12} > 0

$$C_{12} < B < C_{11}; C_{44} > 0$$

Young's modulus (Y)was calculated based on the bulk modulus (B), pressure (C"), hardness (Hu), Kleinman internal strain (ζ), and elastic anisotropy factor (A) of alloys calculated using the relations.

$$Y = 9BG / (3B + G)$$

$$C = C_{12} - C_{44}$$

 $H\upsilon = 0.92 (G/B)^{1.137} G^{0.708}$

$$\zeta = \left(\frac{8C_{12} + C_{11}}{2C_{12} + 7C_{11}}\right)$$
$$A = \left(\frac{C_{44}}{C_{11} - C_{12}}\right)$$

3. Results and discussion

3.1. Structural and electronic properties

The ground state parameters for x = 0.0, 0.25, 0.50, 0.75, and 1 were calculated using the Murnaghan equation [35]. The optimised

lattice was constant for x = 1 and x = 0.0, equal to 4.81 Å and 3.61 Å, respectively, which agrees with other studies [36-48]. The lattice parameters of x = 0.0, 0.25, 0.50, 0.75, and 1 are listed in Table 1. No data about these alloys have been published before. The semiconductor material calculations show that the lattice constant deviation obeys Vegard's law [49-51,12]. This law assumes the location of atoms at particular lattice sites, and the lattice constant varies linearly after inserting the composition x. The calculation results (Fig. 1) show lattice constants vary linearly approximately with x values. Based on the cubic phase, we calculated and analysed the total density of states (DOS) and partial density of states (PDOS) to analyse participation for different levels of B, N, and As atoms. We observed that the structure on the lower-lying energy side of the DOS consisted of a peak located at a close range between 5.5 and 12.5 for all x concentrations. There are other individual peaks due to states of the B 3p, N 4p and As 3 s. The substitution of the N atom by an As atom leads to a change in the As-s position, indicating that part of N-s. DOS and PDOS show three major peaks named VB referring to s, VB' to B-d, and VB" to the B-s and P states. Also, the lower region in the CB due to B-s and P states could be noted. We calculated the E_{g} since this value is critical to better understand the performance of semiconductor optoelectronic devices. The results are listed in Table 2. The band structure calculations of the alloys were implemented after the first relaxation of the positions of atoms. Second, implement with high symmetry positions in the first BZ in cubic phase as presented in Fig. 2. We note that the VBM and CBM for all $BN_{1-x}As_x$ are found at the Γ point, which creates a direct E_g . A direct E_g could be observed in the range of x = 0 to x = 1 of 6.12–1.82 eV, which agrees with other studies as shown in Table 2. When substitution of the N atom by an As atom, the lower part of the VB is dominated by N-d states, and the upper part of the N is dominated by As-p and N-d states, while the lower part of the CB is dominated by B-s and N-p states. Therefore, the $BN_{1-x}As_x$ may be optically active in the visible range. The existence of this property can be considered a desirable factor for using this material in optoelectronic applications.

3.2. Elastic properties

In technological compounds, elastic characteristics play a prominent role. Lagrangian proposed an elasticity theory in which they calculated elastic properties as an anisotropic and homogeneous elastic median. The mechanical characteristics can be determined from elastic constants, which comprise primary data about the natural forces of these compounds. These characteristics then fulfil applications related to mechanical nature, like elastic stress, load deflection, and other applications. The mechanical properties of $BN_{1-x}As_x$ were calculated using Born-Huang (BH) criteria stability. We found that BH easily achieved values for these constants. The BN1-xAs_x alloy was found to be stab, according to the calculated mechanical properties. The Band G values depend on approximating Voigt-Reuss-Hill criteria (VRH). The G and B values are listed in Table 3. The ratio(known as the Pugh ratio (Pr) between G and B for x = 0, 0.5, 0.75, and 1 is less than 1.75, meaning the material is brittle. The Pr value for x = 0.25 is greater than 1.75, so the material is ductile. The calculation result of (V) confirms the brittle property of x = 0, 0.5, 0.75, and 1 due to the V value being lesser than the condition value (0.25), while for x = 0.25 found to be more than 0.25 and confirms ductile property. The C" value determines the ductility and brutality of alloys. The values of C" are negative for alloys where x = 0, 0.5, 0.75, and 1, which confirms their brittle nature. The C" value for x = 0.25 supports ductile nature. The Hv value can be used as a reference for experimental computations due to improvements in G and B macroscopic connotations. For the allows examined, Hu lies within values ranging from 25 to 61 GPa for x = 0, 0.25, 0.5, 0.75, and 1. The ζ value for x = 0, 0.25, 0.5, 0.75, and 1 ranges between 0.37 and 0.61. Due to this range and where ζ does not equal 0, bond bending is reduced, which means the internal displacement within the compounds is small. The A value determines the potential of coaxing microcracks in the alloys. A = 1 for x = 0, 0.25, 0.5, 0.75, and 1; therefore, the nature of alloys is isotropic, but when A not equal to 1 means the alloy is anisotropic.

3.3. Optical properties

Studying the optical properties of semiconductor compounds is vital for practical device systems such as solar cells and photodetectors. The relationship between electronic structure and optical properties of semiconductor alloys was calculated. The calculated

The calculated lattice constants $a_0(A)$ of $BN_{1-x}As_x$ alloy.					
Compositions x	a (Å) This work	a (Å) Previous Studies			
		EXP.	DFT		
0.0	3.61	3.61[40]	3.62[40]		
		4.61[48]	3.615[48]		
		3.616[43]	3.62[43]		
		3.627[41]	3.61[39]		
		3.61[39]	3.62[47]		
			3.61[45]		
0.25	3.95	-	-		
0.5	4.20	-	-		
0.75	4.41	-	-		
1.00	4.81	4.78[42]	4.77[37]		
		4.77[37]	4.77[36]		
		4.77[47]	4.81[47]		

Table 1



Fig. 1. The variation of the calculated lattice constants with change in x for $BN_{1-x}As_x$ alloy.

 Table 2

 The calculated energy gap (Eg) of BN1-xAsx alloy

Compositions x	Eg (eV) This work		Eg (eV) Previous Studies	
	TB-mBJ	PBE-GGA	DFT	EXP.
0.0	6.12	5.95	6.40[40]	6.10[40]
			6.60[41]	5.00[48]
0.25	5.9	5.5	-	-
0.5	3.6	3.2	-	-
0.75	2.4	2.90	-	-
1.0	1.82	1.64	1.69–189[37]	2.02[46]
			1.84[42]	1.82[42]
			1.62[44]	1.77[37]

values for $\epsilon 1(\omega)$ and $\epsilon 2(\omega)$ are shown in Fig. 3. The values of $\epsilon 1(\omega)$ and $\epsilon 2(\omega)$ behaved in the same way when the alloy moved between transitions within x = 0, and x = 1, and this behaves similarly in III-V bonded semiconductor compounds, where the main difference determines the transition energies. The high peak of $\epsilon 1(\omega)$ for x = 0 appears at 9.8 eV, and x = 1 near E \approx 7.3 eV, as shown in Fig. 2(a). When the compound x (0–1) changes, the peaks shift to different energy values, 4.8 eV and 4.7 eV for x = 0.25 and 0.5, respectively. For an allow where x = 0.75, the high value of $\epsilon_1(\omega)$ of 10 eV is due to assigned inter-state transitions. Fig. 2(a) shows $\epsilon_1(\omega)$ for x = 0, where there is a resonant frequency at 10.5 eV; changes in resonant frequencies are found at 4.5 eV for x = 0.25 and x = 0.5 and reaches 10.1 eV for x = 0.75 and 7.5 eV for x = 1. In ϵ^2 (ω), prominent peaks are presented in BN_{1-x}As_x at x = 1, and other x almost have the same peak. They begin from the Γ with type of E₀ transition, which corresponds to standards of E_g and dramatic increase reach to $\epsilon_2 \approx 26$ about 8.5 eV for x = 1. This increase in ϵ_2 case to begin transitions of direct inter-band in Γ with high BZ symmetry. The ϵ_2 value barely shifts to a lower value near 20 eV. This reverted to bands which originated mainly from bands at Γ point. When moving x = 0 to x = 1, all peaks change. The main difference in $\epsilon 2$ (ω) at (x) returns to energies transition. The $\alpha(\omega)$ provides more information about the characteristics of light permeation of particular wavelengths to materials before absorbing light. Also, influences of optical properties on optoelectronic devices concern α (ω). Fig. 2(c) presents this parameter with respect to E_{Ph}. For x = 0 and x = 0.75, we find high peaks of absorption spectra that reach more than 5×10^5 at a range between 11 and 15 eV. When x = 0.25 and 0.5, the high peaks reach more than 3×10^5 at 8 eV, while when x = 1 reach more than 4×10^5 at 6 eV. This means the peaks of absorption spectra varied between x = 0 and x = 1.

The light absorption of x = 0.25, 0.5 and 1 was poor compared with x = 0 and x = 0.75 due to sharing the electrons at the VB edge only in absorption. When E_{ph} increased at x = 0 and x = 0.75, the $\alpha(\omega)$ which led to increased interaction with photons and caused an increase in absorbed photons. Peaks between x = 0 and x = 1 in the low energy region show a red shift. This finding makes $BN_{1-x}As_x$ more important in the short-wave IR range in infrared optoelectronic systems. The $n(\omega)$ values from x = 0 to x = 1 were calculated; The peaks, as shown in Fig. 3(d), constructed from exciting transitions in the E_0 range, could be observed. The peaks vary in intensity between x = 0 and x = 1; the higher peaks are found at x = 1 at 5.1 eV. The significant peak in $n(\omega)$ corresponds to a two-dimensional (2D) excitation transition E_1 . The higher static $n(\omega)$ was calculated to be 3.2 for x = 1, and the lower one is 2.2 at x = 0. The $k(\omega)$ displays a similar trend as $\varepsilon_2(\omega)$. The $k(\omega)$ values are shown in Fig. 2(e). The major peak values for x = 0, 0.25, 0.5, 0.75 and 1 are



Fig. 2. The band structure and density of states of the $BN_{1-x}As_x$ alloy where (a) x = 0, (b) x = 0.25, (c) x = 0.5, (d) x = 0.75 and (e) x = 1.

Table 3

The calculated elastic characteristics of BN1-xAsx alloy.

C ₁₁ 817 2337 997 523 813[48] 680[41]	292 285[42] 275[44]
813[48] 680[41]	285[42] 275[44]
680[41]	275[44]
781[47]	275[37]
798[33]	267[47]
C_{12} 181 1145 420 121	84
	79.5[42]
229[41]	71[44]
190[47]	73[37]
0 171 004 575 004	64[47]
C ₄₄ 4/1 934 5/5 264	155
450[48]	149[42]
	1/0[44]
442[4/]	108[37]
D 272 1542 1071 255	102[47]
B 3/2 1342 10/1 255	132
274_400[42]	140[42]
374-400[40]	139[44]
3/3-402[43]	139[37]
375[30 47]	100[47]
302[45]	
8 0.0026 0.0006 0.0009 0.0	0.005
0.019[41]	0.000
G 399 761 713 234	131
394[48]	128[42]
405-412[43]	138[44]
407[45]	134[47]
E 865 1584 689 477	320
875[48]	326[42]
8554[47]	312[44]
909[45]	300[47]
V 0.12 0.32 0.19 0.1	.8 0.23
0.11[48]	0.22[42]
0.117[5]	0.12[44]
0.07[45]	0.11[47]
Pr = B/G 0.93 2.02 1.50 1.0	1.16
	0.82[42]
	0.96[44]
Hv 60 45 61 40	25
67[48]	
62[45]	
ζ 0.37 0.61 0.55 0.3	.43
0.36[47]	0.4[44]
	0.93[47]
A 1.48 1.56 1.99 1.3	1.5
1.439[47]	1.67[44]
	1.59[47]

observed at 2.6, 2.7, 2.7, 2.3, and 3.7 eV, respectively. Fig. 2(f) represents $L(\omega)$ of electrons passing through the compounds and calculated results for $BN_{1-x}As_x$ for various values of x. The peaks in the $L(\omega)$ curve show characteristics which correspond to plasma correspond for which similar frequency describes plasma frequency, these different match areas of the reflection range. For small peaks, there is a match with the broad plasma resonance at 11.6 eV at lower energies. When x changes, there is a slight change in the plasma frequencies for lower E_{ph} and a slight change in the observation domain. The result is supposed to be the basic standard in optoelectronic devices using semiconductor compounds. The higher peaks are than 40 at 30 eV for x = 0.75, which means the material is more conductive. The reflectivity parameter $R(\omega)$ is considered to be an important parameter in linear optics because of its high sensitivity with respect to groups for $\varepsilon 1(\omega)$ and $\varepsilon_2(\omega)[52,53]$. Fig. 3(g) presents that the upper $R(\omega)$ for x = 0.0, 0.25, 0.5, 0.75 and 1 equals 19, 14, 15, 27, and 10 eV, respectively, which determines the UV spectrum. Fig. 3(h) presented the optical conductivity $\sigma(\omega)$. $\sigma(\omega)$ evolves from zero to a maximum value then decreases dramatically to zero because of the system's metallic character. We find high $\sigma(\omega)$ at 15 eV for x = 0, which agrees with the absorption parameter's peak position. The $\sigma(\omega)$ parameter is a purpose for the VB band transition of As(N)-p with CB of B-s level. The higher $\sigma(\omega)$ is calculated between the visible and UV spectrum.



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Fig. 3. The optical characteristics of the BN_{1-x}As_x where (a) $\varepsilon_1(\omega)$, (b) $\varepsilon_2(\omega)$, (c) $\alpha(\omega)$, (d) $n(\omega)$, (e) $k(\omega)$, (f) $L(\omega)$, (g) $R(\omega)$, and (h) $\sigma(\omega)$.

4. Conclusion

Based on DFT, we demonstrate the structural, electronic, elastic, and optical characteristics of a series of $BN_{1-x}As_x$ alloys of differing compositions. The calculations of lattice parameter values agreed with previous results and were found to undergo slight variations at different concentrations according to Vegard's law. The E_g decreased with an increase in the proportion of As atoms. The elastic parameters of all x = 0, 0.5, 0.75, and 1 showed mechanical stability, and all alloys were brittle except x = 0.25, which was found to be ductile. Our calculations showed that these compounds have small internal displacement due to the ζ value for x = 0, 0.25, 0.5, 0.75, and 1 ranging between 0.37 and 0.61. The (A) values for all the x values were found to be lesser and greater than 1, which indicates anisotropic property. The G and P_r , and other parameters were computed. The main peak of ε_1 (ω) for x = 1 appeared at 7.3 eV, and x = 0 at E .3 s w eV. When the compound x (0-1) changed, peaks shifted at various energies near 5.3 and 5.2 for x = 0.25 and 0.5, respectively, while at a higher value near 10.2 at x = 0.75 due to assigned inter-state transitions. The absorption calculations demonstrate that $BN_{1-x}As_x$ have a higher peak reaching more than 5×10^5 for x = 0.0 and 0.75. The maximum $R(\omega)$ for x = 0.0, 0.25, 0.5, 0.75, and 1 equals 19, 14, 15, 27, and 10 eV, respectively, which determines the UV spectrum. The peaks demonstrate transitions between the VB and CB inter-bands. The calculations of optical coefficients confirm that $BN_{1-x}As_x$ is appropriate for photovoltaic systems. This paper suggests routes for subsequent experimental works in optoelectronic applications.

Declaration of Competing Interest

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests: Moaid K Hussain reports was provided by Alhussain University College. Moaid K Hussain reports a relationship with Alhussain University College that includes: employment. Moaid K Hussain has patent pending to Optik. Hayder Salah Mohammed.

Data Availability

No data was used for the research described in the article.

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